

First Principles Study of Small Hole Polaron Formation in Doped Olivine $\text{LiFe}_{1-x}\text{Co}_x\text{PO}_4$: Effects of Li Deficiency

การศึกษาการเกิดโฮลโพลารอนขนาดเล็กในสารโอลิวีนที่ถูกเจือ ($\text{LiFe}_{1-x}\text{Co}_x\text{PO}_4$)

โดยมีผลมาจากการดัดแปลงเชื่อมออกด้วยวิธีคำนวณแบบเฟิสต์พริ้นซิเพิล

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Understanding the localization of small polarons (p^+) are important for improvement of electronic conductivity in cathode materials such as LiFePO_4 and LiCoPO_4 for Li-ion battery technology. In this study we calculate the formation energy (ΔF) due to Li vacancies in $\text{LiFe}_{1-x}\text{Co}_x\text{PO}_4$ by density functional theory with additional on-site Hubbard correction (DFT+U). We found that Li vacancies (V_{Li^-}) can induce polaron formation preferentially at Fe rather than Co sites. Co also improves polaron binding to V_{Li^-} to form a small polaron complex ($V_{\text{Li}^-} - p^+$) and alters $\text{Li}_{1-y}\text{FePO}_4$ to the equilibrium phase ($\Delta F < 0$). Co doping can reduce the energy gap and increase the intercalation voltage. Overall, polaron localization plays key roles in electronic conductivity of $\text{LiFe}_{1-x}\text{Co}_x\text{PO}_4$ cathode materials.

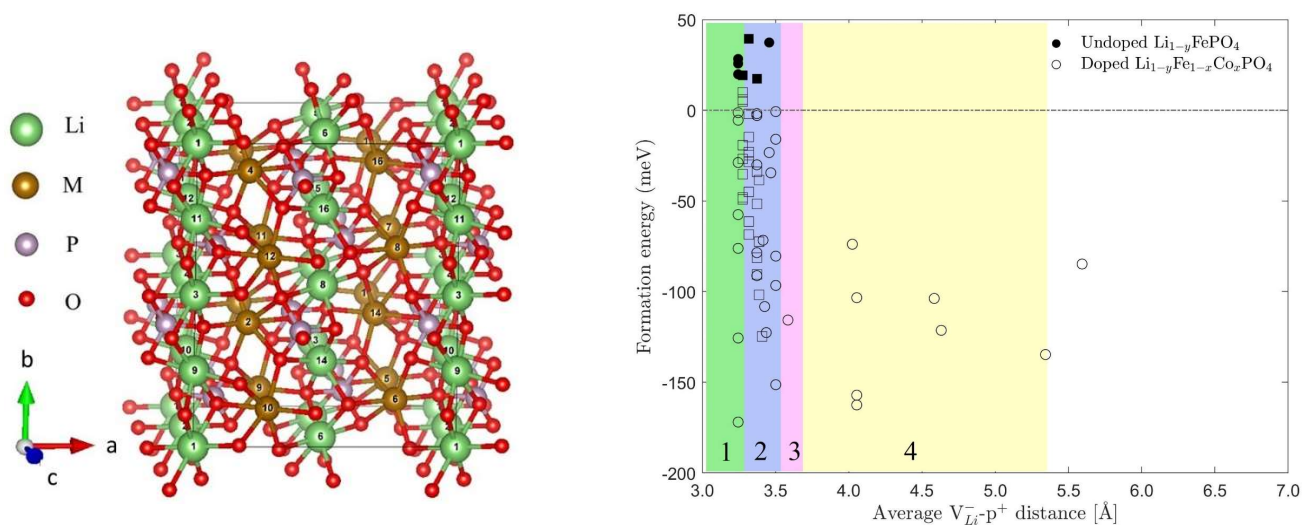


Figure : (Left) Olivine-type LiMPO_4 ($M=\text{Fe}, \text{Co}$) structure and (Right) formation energy (ΔF) of a small polaron complex ($V_{\text{Li}^-} - p^+$) in an equilibrium position as a function of distance between V_{Li^-} and p^+ where the circle and square symbols denote $y \leq 0.50$ and $y > 0.50$ respectively. The shaded regions with the number represented for each nearest neighbor (NN). 1=1st NN, 2=2nd NN, 3=3rd NN and 4=4th NN.